

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

2-(4-Fluorophenyl)-N-{4-[6-(4-fluorophenyl)-2,3-dihydroimidazo[2,1-b]-[1,3]thiazol-5-yl]pyridin-2-yl}acetamide

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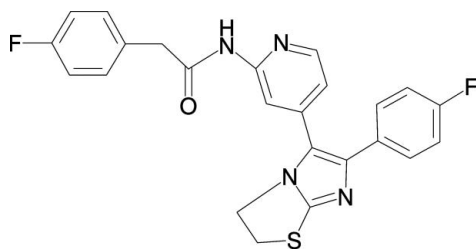
Received 16 March 2010; accepted 6 April 2010

Key indicators: single-crystal X-ray study; $T = 173$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; disorder in main residue; R factor = 0.038; wR factor = 0.090; data-to-parameter ratio = 16.3.

In the crystal structure of the title compound, $\text{C}_{24}\text{H}_{18}\text{F}_2\text{N}_4\text{OS}$, the imidazole system makes dihedral angles of 34.3 (1) and 43.9 (1)°, respectively, with the directly attached 4-fluorophenyl and pyridine rings. The crystal structure is stabilized by intermolecular $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonding and by an intramolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen interaction. The F atom of the 2-(4-fluorophenyl) group is disordered over two positions with site-occupancy factors of 0.75 and 0.25.

Related literature

For related compounds and their biological relevance, see: Ziegler *et al.* (2009).



Experimental

Crystal data

$\text{C}_{24}\text{H}_{18}\text{F}_2\text{N}_4\text{OS}$

$M_r = 448.48$

Monoclinic, Cc
 $a = 4.9179$ (3) Å
 $b = 23.592$ (1) Å
 $c = 18.4834$ (9) Å
 $\beta = 91.523$ (2)°
 $V = 2143.8$ (2) Å³

$Z = 4$
Mo $K\alpha$ radiation
 $\mu = 0.19$ mm⁻¹
 $T = 173$ K
 $0.35 \times 0.16 \times 0.08$ mm

Data collection

Bruker SMART APEXII
diffractometer
10277 measured reflections

4846 independent reflections
4129 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.028$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.038$
 $wR(F^2) = 0.090$
 $S = 1.03$
4846 reflections
298 parameters
2 restraints

H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.23$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.20$ e Å⁻³
Absolute structure: Flack (1983),
2197 Friedel pairs
Flack parameter: 0.07 (6)

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N13}-\text{H13}\cdots\text{N6}^i$	0.97	2.02	2.980 (2)	171
$\text{C8}-\text{H8}\cdots\text{O15}$	0.95	2.24	2.845 (3)	120

Symmetry code: (i) $x - \frac{1}{2}, -y + \frac{1}{2}, z + \frac{1}{2}$.

Data collection: *APEX2* (Bruker, 2006); cell refinement: *SAINTE* (Bruker, 2006); data reduction: *SAINTE*; program(s) used to solve structure: *SIR97* (Altomare *et al.*, 1999); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *PLATON*.

The authors would like to thank the Federal Ministry of Education and Research, Germany, Merckle GmbH, Ulm, Germany, and Fonds der Chemischen Industrie, Germany, for their generous support of this work.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: IM2189).

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supplementary materials

Acta Cryst. (2010). E66, o1132 [doi:10.1107/S1600536810012766]

2-(4-Fluorophenyl)-*N*-{4-[6-(4-fluorophenyl)-2,3-dihydroimidazo[2,1-*b*][1,3]thiazol-5-yl]pyridin-2-yl}acetamide

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Comment

SKF86002 was an early lead compound of many imidazole based p38 MAP kinase inhibitors. A further improvement of these inhibitors was the modification of the pyridyl moiety by substitution with amines in C-2 position of the pyridine. This donor acceptor system can interact with hinge region/MET109 by bidentate hydrogen bonding. Additional interactions of the attached residues with the hydrophobic region II strongly increases potency of these compounds (Ziegler *et al.* 2009).

The imidazole system of the title compound 2-(4-fluorophenyl)-*N*-{4-[6-(4-fluorophenyl)-2,3-dihydroimidazo[2,1-*b*][1,3]thiazol-5-yl]-pyridin-2-yl}acetamide, C₂₄H₁₈F₂N₄OS, makes dihedral angles of 34.3 (1)° and 43.9 (1)° with the directly attached 4-fluorophenyl and the pyridine rings, respectively. The crystal structure is characterized by an intermolecular hydrogen bond N13—H13⋯N6 (2.02 Å). The molecular conformation is stabilized by an intramolecular C8—H8⋯O15 (2.24 Å) interaction. The fluorine atom F1 is disordered over two positions with s.o.f 0.75:0.25.

Experimental

4-fluorophenylacetic acid (296 mg) was dissolved in 3 ml *N*-methylpyrrolidinone. After addition of 332 mg of carbonyldiimidazole the mixture was stirred for 1 h at room temperature. 200 mg 4-[6-(4-fluorophenyl)-2,3-dihydroimidazo[2,1-*b*][1,3]thiazol-5-yl]pyridin-2-amine was added and the reaction mixture was heated to 383 K for 19 h. The reaction mixture was quenched with a solution of concentrated sodium hydrogen carbonate and extracted with ethylacetate. The crude product was purified by flash chromatography (eluent: ethylacetate/hexane 2/1) to yield 155 mg (54%) of the title compound. Crystals suitable for X-ray analysis were obtained by slow crystallization from methanol at room temperature.

Refinement

Hydrogen atoms attached to carbons were placed at calculated positions with C—H = 0.95 Å (aromatic) or 0.98–0.99 Å (*sp*³ C-atom). All H atoms were refined in the riding-model approximation with isotropic displacement parameters (set at 1.2–1.5 times of the *U*_{eq} of the parent atom). The fluorine F1 is disordered over two positions with s.o.f 0.75:0.25.

Figures

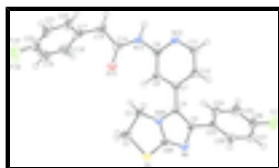


Fig. 1. View of compound I. Displacement ellipsoids are drawn at the 50% probability level.

2-(4-Fluorophenyl)-N-{4-[6-(4-fluorophenyl)-2,3-dihydroimidazo[2,1-b][1,3]thiazol-5-yl]pyridin-2-yl}acetamide

Crystal data

$C_{24}H_{18}F_2N_4OS$	$F(000) = 928$
$M_r = 448.48$	$D_x = 1.390 \text{ Mg m}^{-3}$
Monoclinic, Cc	Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
Hall symbol: C -2yc	Cell parameters from 3757 reflections
$a = 4.9179 (3) \text{ \AA}$	$\theta = 2.2\text{--}26.4^\circ$
$b = 23.592 (1) \text{ \AA}$	$\mu = 0.19 \text{ mm}^{-1}$
$c = 18.4834 (9) \text{ \AA}$	$T = 173 \text{ K}$
$\beta = 91.523 (2)^\circ$	Plate, yellow
$V = 2143.8 (2) \text{ \AA}^3$	$0.35 \times 0.16 \times 0.08 \text{ mm}$
$Z = 4$	

Data collection

Bruker SMART APEXII diffractometer	4129 reflections with $I > 2\sigma(I)$
Radiation source: sealed tube graphite	$R_{\text{int}} = 0.028$
CCD scan	$\theta_{\text{max}} = 28.2^\circ$, $\theta_{\text{min}} = 1.7^\circ$
10277 measured reflections	$h = -6 \rightarrow 6$
4846 independent reflections	$k = -30 \rightarrow 28$
	$l = -23 \rightarrow 24$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.038$	H-atom parameters constrained
$wR(F^2) = 0.090$	$w = 1/[\sigma^2(F_o^2) + (0.0433P)^2 + 0.4891P]$
$S = 1.02$	where $P = (F_o^2 + 2F_c^2)/3$
4846 reflections	$(\Delta/\sigma)_{\text{max}} = 0.001$
298 parameters	$\Delta\rho_{\text{max}} = 0.23 \text{ e \AA}^{-3}$
2 restraints	$\Delta\rho_{\text{min}} = -0.20 \text{ e \AA}^{-3}$
Primary atom site location: structure-invariant direct methods	Absolute structure: Flack (1983), 2197 Friedel pairs
	Flack parameter: 0.07 (6)

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
S1	0.82180 (11)	0.16231 (2)	-0.12400 (3)	0.03583 (14)	
F1A	-0.3136 (9)	-0.09848 (19)	0.2022 (2)	0.0837 (13)	0.75
F1B	-0.454 (3)	-0.0837 (5)	0.1902 (9)	0.086 (4)	0.25
F2	0.7391 (4)	0.53748 (6)	0.01213 (10)	0.0609 (4)	
C2	0.6645 (6)	0.11864 (10)	-0.05475 (12)	0.0420 (6)	
H2A	0.4683	0.1135	-0.0662	0.050*	
H2B	0.7517	0.0808	-0.0525	0.050*	
C3	0.7039 (6)	0.14935 (9)	0.01721 (11)	0.0386 (6)	
H3A	0.5578	0.1394	0.0508	0.046*	
H3B	0.8821	0.1398	0.0402	0.046*	
N3A	0.6911 (4)	0.20911 (7)	-0.00241 (9)	0.0294 (4)	
C4	0.6770 (4)	0.26014 (9)	0.03519 (10)	0.0277 (4)	
C5	0.7275 (4)	0.30126 (9)	-0.01557 (10)	0.0266 (4)	
N6	0.7779 (4)	0.27699 (7)	-0.08266 (8)	0.0302 (4)	
C6A	0.7576 (5)	0.22229 (9)	-0.07094 (10)	0.0301 (4)	
C7	0.6335 (4)	0.26246 (8)	0.11346 (10)	0.0264 (4)	
C8	0.4394 (4)	0.22883 (9)	0.14539 (10)	0.0265 (4)	
H8	0.3224	0.2053	0.1167	0.032*	
C9	0.4195 (4)	0.23019 (9)	0.22064 (10)	0.0251 (4)	
N10	0.5728 (4)	0.26393 (7)	0.26344 (9)	0.0289 (4)	
C11	0.7533 (5)	0.29740 (9)	0.23135 (11)	0.0313 (5)	
H11	0.8601	0.3222	0.2610	0.038*	
C12	0.7934 (4)	0.29797 (9)	0.15769 (10)	0.0281 (4)	
H12	0.9266	0.3220	0.1375	0.034*	
N13	0.2423 (4)	0.19595 (7)	0.25989 (8)	0.0269 (4)	
H13	0.2720	0.2071	0.3102	0.032*	
C14	0.0701 (4)	0.15550 (9)	0.23346 (11)	0.0289 (4)	
O15	0.0457 (4)	0.14302 (8)	0.16963 (8)	0.0429 (4)	
C16	-0.0944 (5)	0.12612 (10)	0.29138 (11)	0.0328 (5)	
H16A	0.0147	0.1247	0.3371	0.039*	
H16B	-0.2605	0.1485	0.3002	0.039*	
C17	-0.1743 (5)	0.06690 (10)	0.26956 (11)	0.0339 (5)	
C18	-0.3773 (6)	0.05731 (15)	0.21804 (14)	0.0533 (7)	
H18	-0.4780	0.0881	0.1979	0.064*	
C19	-0.4342 (8)	0.00127 (19)	0.19545 (17)	0.0757 (12)	
H19	-0.5744	-0.0061	0.1604	0.091*	
C20	-0.2856 (9)	-0.04181 (15)	0.22459 (18)	0.0753 (12)	
C21	-0.0855 (9)	-0.03416 (13)	0.27436 (18)	0.0693 (10)	
H21	0.0149	-0.0654	0.2934	0.083*	
C22	-0.0297 (6)	0.02066 (11)	0.29714 (14)	0.0496 (7)	

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H22	0.1109	0.0268	0.3325	0.060*
C23	0.7289 (4)	0.36344 (8)	-0.00730 (10)	0.0254 (4)
C24	0.5520 (5)	0.39102 (9)	0.03857 (11)	0.0312 (5)
H24	0.4270	0.3695	0.0656	0.037*
C25	0.5560 (5)	0.44968 (10)	0.04538 (12)	0.0379 (5)
H25	0.4361	0.4684	0.0770	0.045*
C26	0.7373 (5)	0.47988 (9)	0.00533 (12)	0.0371 (5)
C27	0.9136 (5)	0.45469 (10)	-0.04073 (12)	0.0383 (5)
H27	1.0359	0.4768	-0.0680	0.046*
C28	0.9095 (5)	0.39598 (9)	-0.04675 (11)	0.0320 (5)
H28	1.0316	0.3777	-0.0782	0.038*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0618 (4)	0.0249 (2)	0.0210 (2)	-0.0009 (3)	0.0078 (2)	-0.0022 (2)
F1A	0.123 (4)	0.061 (3)	0.068 (2)	-0.052 (2)	0.019 (2)	-0.0149 (17)
F1B	0.110 (10)	0.033 (5)	0.113 (9)	-0.036 (6)	-0.005 (8)	-0.016 (5)
F2	0.0845 (12)	0.0243 (7)	0.0745 (11)	-0.0014 (8)	0.0124 (9)	-0.0004 (7)
C2	0.0714 (18)	0.0268 (11)	0.0284 (11)	-0.0066 (11)	0.0095 (11)	0.0002 (9)
C3	0.0685 (17)	0.0239 (11)	0.0234 (10)	-0.0072 (11)	0.0050 (10)	0.0033 (8)
N3A	0.0477 (11)	0.0230 (8)	0.0177 (7)	-0.0027 (8)	0.0049 (7)	0.0016 (6)
C4	0.0364 (11)	0.0260 (10)	0.0209 (9)	-0.0048 (9)	0.0041 (8)	0.0005 (7)
C5	0.0344 (11)	0.0265 (10)	0.0191 (9)	-0.0031 (9)	0.0051 (8)	-0.0002 (7)
N6	0.0458 (11)	0.0254 (9)	0.0196 (8)	-0.0018 (8)	0.0061 (7)	0.0011 (7)
C6A	0.0454 (12)	0.0294 (11)	0.0158 (8)	-0.0029 (10)	0.0046 (8)	-0.0009 (8)
C7	0.0362 (11)	0.0262 (11)	0.0170 (9)	0.0029 (9)	0.0032 (8)	0.0011 (7)
C8	0.0326 (11)	0.0267 (10)	0.0201 (9)	-0.0020 (9)	0.0025 (8)	-0.0009 (7)
C9	0.0321 (11)	0.0240 (10)	0.0194 (9)	0.0031 (8)	0.0056 (8)	0.0002 (7)
N10	0.0414 (11)	0.0277 (9)	0.0179 (7)	-0.0015 (8)	0.0048 (7)	-0.0015 (6)
C11	0.0441 (13)	0.0263 (10)	0.0235 (9)	-0.0045 (9)	0.0013 (9)	-0.0043 (8)
C12	0.0356 (11)	0.0241 (10)	0.0249 (9)	-0.0017 (9)	0.0046 (8)	0.0001 (8)
N13	0.0374 (10)	0.0265 (9)	0.0172 (7)	-0.0021 (8)	0.0071 (7)	0.0009 (6)
C14	0.0314 (11)	0.0313 (11)	0.0240 (9)	0.0022 (9)	0.0015 (8)	0.0049 (8)
O15	0.0533 (10)	0.0536 (11)	0.0219 (7)	-0.0213 (8)	0.0002 (7)	0.0032 (7)
C16	0.0371 (12)	0.0358 (12)	0.0259 (10)	-0.0019 (9)	0.0075 (8)	0.0047 (8)
C17	0.0352 (12)	0.0410 (13)	0.0260 (9)	-0.0113 (10)	0.0089 (8)	0.0041 (9)
C18	0.0392 (14)	0.080 (2)	0.0405 (14)	-0.0095 (14)	0.0002 (11)	-0.0043 (13)
C19	0.069 (2)	0.118 (3)	0.0396 (15)	-0.054 (2)	0.0048 (15)	-0.0204 (18)
C20	0.117 (3)	0.058 (2)	0.0525 (18)	-0.054 (2)	0.027 (2)	-0.0082 (15)
C21	0.112 (3)	0.0334 (15)	0.0633 (18)	-0.0212 (16)	0.0105 (19)	0.0129 (13)
C22	0.0641 (18)	0.0387 (14)	0.0457 (14)	-0.0164 (13)	-0.0073 (12)	0.0132 (11)
C23	0.0330 (10)	0.0254 (10)	0.0177 (8)	-0.0011 (9)	-0.0014 (7)	0.0018 (7)
C24	0.0357 (12)	0.0290 (11)	0.0291 (10)	-0.0002 (9)	0.0056 (9)	0.0063 (8)
C25	0.0435 (13)	0.0351 (12)	0.0352 (12)	0.0086 (10)	0.0053 (10)	0.0003 (9)
C26	0.0521 (14)	0.0205 (10)	0.0384 (12)	-0.0024 (10)	-0.0036 (10)	0.0018 (9)
C27	0.0468 (14)	0.0334 (12)	0.0348 (11)	-0.0126 (10)	0.0026 (10)	0.0074 (9)
C28	0.0381 (13)	0.0320 (12)	0.0263 (10)	-0.0046 (9)	0.0058 (9)	-0.0004 (8)

Geometric parameters (Å, °)

S1—C6A	1.755 (2)	N13—C14	1.358 (3)
S1—C2	1.831 (2)	N13—H13	0.9732
F1A—C20	1.405 (5)	C14—O15	1.219 (2)
F1B—C20	1.429 (13)	C14—C16	1.526 (3)
F2—C26	1.365 (2)	C16—C17	1.503 (3)
C2—C3	1.522 (3)	C16—H16A	0.9900
C2—H2A	0.9900	C16—H16B	0.9900
C2—H2B	0.9900	C17—C18	1.380 (3)
C3—N3A	1.457 (3)	C17—C22	1.391 (4)
C3—H3A	0.9900	C18—C19	1.412 (5)
C3—H3B	0.9900	C18—H18	0.9500
N3A—C6A	1.353 (2)	C19—C20	1.355 (6)
N3A—C4	1.393 (3)	C19—H19	0.9500
C4—C5	1.377 (3)	C20—C21	1.341 (5)
C4—C7	1.469 (3)	C21—C22	1.385 (4)
C5—N6	1.394 (2)	C21—H21	0.9500
C5—C23	1.475 (3)	C22—H22	0.9500
N6—C6A	1.313 (3)	C23—C24	1.392 (3)
C7—C8	1.385 (3)	C23—C28	1.394 (3)
C7—C12	1.398 (3)	C24—C25	1.390 (3)
C8—C9	1.397 (3)	C24—H24	0.9500
C8—H8	0.9500	C25—C26	1.373 (3)
C9—N10	1.340 (3)	C25—H25	0.9500
C9—N13	1.405 (3)	C26—C27	1.367 (4)
N10—C11	1.338 (3)	C27—C28	1.390 (3)
C11—C12	1.381 (3)	C27—H27	0.9500
C11—H11	0.9500	C28—H28	0.9500
C12—H12	0.9500		
C6A—S1—C2	88.70 (10)	O15—C14—C16	121.9 (2)
C3—C2—S1	107.26 (15)	N13—C14—C16	113.82 (17)
C3—C2—H2A	110.3	C17—C16—C14	111.93 (17)
S1—C2—H2A	110.3	C17—C16—H16A	109.2
C3—C2—H2B	110.3	C14—C16—H16A	109.2
S1—C2—H2B	110.3	C17—C16—H16B	109.2
H2A—C2—H2B	108.5	C14—C16—H16B	109.2
N3A—C3—C2	103.86 (16)	H16A—C16—H16B	107.9
N3A—C3—H3A	111.0	C18—C17—C22	118.5 (3)
C2—C3—H3A	111.0	C18—C17—C16	121.1 (2)
N3A—C3—H3B	111.0	C22—C17—C16	120.2 (2)
C2—C3—H3B	111.0	C17—C18—C19	119.4 (3)
H3A—C3—H3B	109.0	C17—C18—H18	120.3
C6A—N3A—C4	106.57 (16)	C19—C18—H18	120.3
C6A—N3A—C3	116.46 (17)	C20—C19—C18	119.0 (3)
C4—N3A—C3	135.62 (17)	C20—C19—H19	120.5
C5—C4—N3A	104.88 (17)	C18—C19—H19	120.5
C5—C4—C7	132.72 (19)	C21—C20—C19	123.3 (3)

supplementary materials

N3A—C4—C7	122.28 (17)	C21—C20—F1A	113.2 (4)
C4—C5—N6	110.87 (18)	C19—C20—F1A	123.3 (4)
C4—C5—C23	129.10 (17)	C21—C20—F1B	143.6 (7)
N6—C5—C23	120.02 (16)	C19—C20—F1B	92.4 (6)
C6A—N6—C5	103.95 (15)	C20—C21—C22	118.0 (3)
N6—C6A—N3A	113.68 (17)	C20—C21—H21	121.0
N6—C6A—S1	133.22 (15)	C22—C21—H21	121.0
N3A—C6A—S1	112.93 (15)	C21—C22—C17	121.7 (3)
C8—C7—C12	118.51 (17)	C21—C22—H22	119.2
C8—C7—C4	121.22 (18)	C17—C22—H22	119.2
C12—C7—C4	120.25 (18)	C24—C23—C28	118.57 (19)
C7—C8—C9	118.56 (18)	C24—C23—C5	121.78 (19)
C7—C8—H8	120.7	C28—C23—C5	119.65 (19)
C9—C8—H8	120.7	C25—C24—C23	120.9 (2)
N10—C9—C8	123.24 (19)	C25—C24—H24	119.6
N10—C9—N13	112.57 (16)	C23—C24—H24	119.6
C8—C9—N13	124.18 (18)	C26—C25—C24	118.4 (2)
C11—N10—C9	117.27 (16)	C26—C25—H25	120.8
N10—C11—C12	123.80 (19)	C24—C25—H25	120.8
N10—C11—H11	118.1	F2—C26—C27	119.2 (2)
C12—C11—H11	118.1	F2—C26—C25	118.0 (2)
C11—C12—C7	118.56 (19)	C27—C26—C25	122.8 (2)
C11—C12—H12	120.7	C26—C27—C28	118.4 (2)
C7—C12—H12	120.7	C26—C27—H27	120.8
C14—N13—C9	127.44 (16)	C28—C27—H27	120.8
C14—N13—H13	127.6	C27—C28—C23	121.0 (2)
C9—N13—H13	105.0	C27—C28—H28	119.5
O15—C14—N13	124.24 (19)	C23—C28—H28	119.5
C6A—S1—C2—C3	-27.8 (2)	C4—C7—C12—C11	-177.6 (2)
S1—C2—C3—N3A	33.4 (2)	N10—C9—N13—C14	-176.67 (19)
C2—C3—N3A—C6A	-25.1 (3)	C8—C9—N13—C14	2.2 (3)
C2—C3—N3A—C4	170.3 (2)	C9—N13—C14—O15	0.0 (4)
C6A—N3A—C4—C5	2.2 (2)	C9—N13—C14—C16	-179.99 (19)
C3—N3A—C4—C5	167.9 (3)	O15—C14—C16—C17	27.2 (3)
C6A—N3A—C4—C7	-174.4 (2)	N13—C14—C16—C17	-152.78 (19)
C3—N3A—C4—C7	-8.7 (4)	C14—C16—C17—C18	-74.3 (3)
N3A—C4—C5—N6	-1.3 (2)	C14—C16—C17—C22	100.8 (2)
C7—C4—C5—N6	174.8 (2)	C22—C17—C18—C19	0.6 (4)
N3A—C4—C5—C23	177.6 (2)	C16—C17—C18—C19	175.8 (2)
C7—C4—C5—C23	-6.3 (4)	C17—C18—C19—C20	-0.6 (4)
C4—C5—N6—C6A	-0.2 (3)	C18—C19—C20—C21	0.1 (5)
C23—C5—N6—C6A	-179.2 (2)	C18—C19—C20—F1A	-175.5 (3)
C5—N6—C6A—N3A	1.7 (3)	C18—C19—C20—F1B	172.8 (7)
C5—N6—C6A—S1	-173.1 (2)	C19—C20—C21—C22	0.3 (5)
C4—N3A—C6A—N6	-2.5 (3)	F1A—C20—C21—C22	176.3 (3)
C3—N3A—C6A—N6	-171.4 (2)	F1B—C20—C21—C22	-167.3 (12)
C4—N3A—C6A—S1	173.32 (16)	C20—C21—C22—C17	-0.2 (5)
C3—N3A—C6A—S1	4.5 (3)	C18—C17—C22—C21	-0.2 (4)
C2—S1—C6A—N6	-170.9 (3)	C16—C17—C22—C21	-175.4 (3)

C2—S1—C6A—N3A	14.27 (19)	C4—C5—C23—C24	-34.4 (3)
C5—C4—C7—C8	140.0 (2)	N6—C5—C23—C24	144.5 (2)
N3A—C4—C7—C8	-44.5 (3)	C4—C5—C23—C28	146.3 (2)
C5—C4—C7—C12	-41.6 (4)	N6—C5—C23—C28	-34.9 (3)
N3A—C4—C7—C12	133.9 (2)	C28—C23—C24—C25	-0.4 (3)
C12—C7—C8—C9	-2.5 (3)	C5—C23—C24—C25	-179.7 (2)
C4—C7—C8—C9	175.9 (2)	C23—C24—C25—C26	0.5 (3)
C7—C8—C9—N10	2.4 (3)	C24—C25—C26—F2	179.8 (2)
C7—C8—C9—N13	-176.35 (19)	C24—C25—C26—C27	0.0 (4)
C8—C9—N10—C11	-0.3 (3)	F2—C26—C27—C28	179.8 (2)
N13—C9—N10—C11	178.53 (19)	C25—C26—C27—C28	-0.4 (4)
C9—N10—C11—C12	-1.5 (3)	C26—C27—C28—C23	0.5 (3)
N10—C11—C12—C7	1.3 (3)	C24—C23—C28—C27	-0.1 (3)
C8—C7—C12—C11	0.9 (3)	C5—C23—C28—C27	179.3 (2)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
N13—H13 \cdots N6 ⁱ	0.97	2.02	2.980 (2)	171
C8—H8 \cdots O15	0.95	2.24	2.845 (3)	120

Symmetry codes: (i) $x-1/2, -y+1/2, z+1/2$.

Fig. 1

